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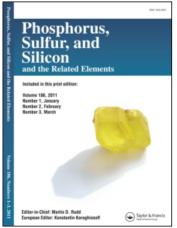
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## Phosphorus, Sulfur, and Silicon and the Related Elements

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## 1,3-Bisylidylsubstituted Cyclotetraphosphines

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## 1,3-Bisylidylsubstituted Cyclotetraphosphines

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Ylide substituents effectively stabilize low coordination of phosphorus as long as it is bonded to electronegative partners<sup>1,2</sup>. Thus ylidyl-thioxophosphines and -selenoxophosphines are monomeric in solution and crystal<sup>3</sup>. In contrast ylidsubstituted disphosphines are unstable towards dimerization in accord with the anticipated polarity of the PP-bond:

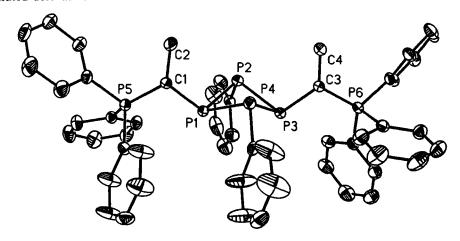
Consequently, the reaction of ylidyldichlorophosphines and bis(trimethylsilyl)phenylphosphine yields cyclotetraphosphines:

<sup>31</sup>P-NMR Data of the cyclotetraphosphines

R 
$$\delta_{A}$$
  $\delta_{B}$   $\delta_{C}$   $^{6}J_{AA}$ ,  $^{2}J_{AB}$   $^{4}J_{AB}$ ,  $^{3}J_{AC}$   $^{2}J_{BB}$ ,  $^{1}J_{BC}$   
Me 23.8 -13.6 -44.0 2.0 163.2 2.9 6.2 150.0 -129.9 3-MeC<sub>6</sub>H<sub>4</sub> 19.4 -22.0 -29.1 2.9 183.2 3.0 5.6 179.7 -131.8

Their  $^{31}$ P-NMR spectra represent AA'BB'C<sub>2</sub>-spinsystems indicating the expected head/tail-dimer in an all-trans configuration. The two-bond PP-coupling constants  $J_{\rm BB}$ , are exceptionally high and point to a transanular approach of the lone pairs at the two concerned phosphorus ring members. A similarly strong coupling was observed before only for a head/head-dimer<sup>4</sup>.

These conclusions are confirmed by the X-ray structure analysis of the methyl-substituted derivative:



Molecular structure of the methyl derivative (without hydrogen atoms, thermal ellipsoids at the 25% probability level)

The tetraphosphine ring is folded, with all substituents in equatorial positions. The ylide carbon atoms are (almost) planar coordinated (sum of angles 360° and 356°). As it is generally the case in C-phosphino-substituted phosphonium ylides the phosphoniogroups are orientated synperiplanar to the electron lone pairs<sup>5,6</sup> at P1 and P3. The structure seems to strive for a relatively long distance between the phosphonium centers P5 and P6: The tetraphosphetane ring is less strongly folded along P2···P4 than along P1···P3 (interplanar angles 139.8° and 122.5°, respectively) and the pyramids at P1 and P3 are less steep than those at the phenyl-substituted atoms P2 and P4 (averaged sum of angles 299° and 289°, respectively).

- 1. A. Schmidpeter and G. Jochem, Tetrahedron Lett. 1992, 33, 471 474.
- A. Schmidpeter, G. Jochem, K. Karaghiosoff and Ch. Robl, Angew. Chem. 1992, 104, 1420 - 1421; Angew. Chem. Int. Ed. Engl. 1992, 31, 1350 - 1352.
- G. Jochem, H. Nöth and A. Schmidpeter, Angew. Chem. 1993, 105, 1117 1119;
   Angew. Chem. Int. Ed. Engl. 1993, 32, 1089 1091.
- 4. R. Appel, D. Gudat, E. Niecke, M. Nieger, C. Porz and H. Westermann, Z. Naturforsch. B 1991, 46, 865 883.
- 5. H. Schmidbaur, U. Deschler, B. Milewski-Mahrla, Chem. Ber. 1983, 116, 1393-1402.
- 6. A. Schmidpeter, H. Nöth, G. Jochem, H.-P. Schrödel and K. Karaghiosoff, *Chem. Ber.*, submitted for publication.